

DEPTH OF BIOLOGY

B. PHARMACY

7 SEM IMPORTANT QUESTIONS

**INSTRUMENTAL METHOD OF
ANALYSIS**

UNIT –I

10 Hours

UV Visible spectroscopy

Electronic transitions, chromophores, auxochromes, spectral shifts, solvent effect on absorption spectra, Beer and Lambert's law, Derivation and deviations.

Instrumentation - Sources of radiation, wavelength selectors, sample cells, detectors- Photo tube, Photomultiplier tube, Photo voltaic cell, Silicon Photodiode.

Applications - Spectrophotometric titrations, Single component and multi component analysis

Fluorimetry

Theory, Concepts of singlet, doublet and triplet electronic states, internal and external conversions, factors affecting fluorescence, quenching, instrumentation and applications

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10 MARKS

Q1. Describe various Electronic transition with the help of energy orbital diagram ?

Q2. Describe fluorimetry it's Theory , concept and factor affecting fluorescence?

Q3. Explain spectrophotometry Titration and also describe single / multicomponent analysis .

5 MARKS

Q1. Explain Beer and Lambert 's law (derivation)

Q2. Explain fluorimetry and factor affecting fluorescence.

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Q3. Describe the effect of solvent on absorption spectra ?

2 MARKS

Q1. Explain Beer's Law .

Q2. Explain Lambert's Law .

Q3. What is chromophores ?

Q4. Define Auxochromes ?

Q5. Role of silicon photodiode ?

Q6. Define Fluorimetry ?

Q7. What is PMT in UV - Visible spectroscops ?

Q8. Define Quenching ?

UNIT –II

10 Hours

IR spectroscopy

Introduction, fundamental modes of vibrations in poly atomic molecules, sample handling, factors affecting vibrations

Instrumentation - Sources of radiation, wavelength selectors, detectors - Golay cell, Bolometer, Thermocouple, Thermister, Pyroelectric detector and applications

Flame Photometry-Principle, interferences, instrumentation and applications

Atomic absorption spectroscopy- Principle, interferences, instrumentation and applications

Nepheloturbidometry- Principle, instrumentation and applications

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10 MARKS

Q1. Describe I.R. Spectroscopy and also explain fundamental mode of vibration in polyatomic molecule and its factor affecting .

Q2. Give a detail note on principle , Instrumentation and application of AAS and Nepheloturbidometry .

5 MARKS

Q1. Explain Principle , Instrument and application of Flame Photometry / Nepheloturbidometry .

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2 MARKS

Q1. Explain IR Spectroscopy .

Q2. Define sample handling

Q3. What are the various factor which affect the vibrational frequency in I.R.

Q4. Define Flame Photometry ?

Q5. Write down the principle of Atomic Absorbption spectroscopy (AAS) ?

Q6. Define Nepheloturbidometry ?

Q7. Name the diluents used for solid - sample in IR - Spectroscopy ?

UNIT –III

10 Hours

Introduction to chromatography

Adsorption and partition column chromatography-Methodology, advantages, disadvantages and applications.

Thin layer chromatography- Introduction, Principle, Methodology, Rf values, advantages, disadvantages and applications.

Paper chromatography-Introduction, methodology, development techniques, advantages, disadvantages and applications

Electrophoresis– Introduction, factors affecting electrophoretic mobility, Techniques of paper, gel, capillary electrophoresis, applications

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10 MARKS

Q1. Define Electrophoresis it's factor affecting and application .

Q2. Explain different type of chromatography (Gas , Thin layer , paper and partition Column) .

5 MARKS

Q1. Give a detailed account on paper chromatography ?

Q2. Give a detailed note on partition Column Chromatography ?

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2 MARKS

Q1. Define chromatography ?

Q2. Explain Electrophoresis .

Q3. What are the application of chromatography .

Q4. Define Adsorption and Partition Column Chromatography ?

Q5. Define R.F. Values .

UNIT –IV

08 Hours

Gas chromatography - Introduction, theory, instrumentation, derivatization, temperature programming, advantages, disadvantages and applications

High performance liquid chromatography (HPLC)-Introduction, theory, instrumentation, advantages and applications.

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10 MARKS

Q1. Give a detail note on Gas Chromatography .

Q2. Explain HPLC it's theory Instrumentation and application .

5 MARKS

Q1. Write a note on HPLC ?

Q2. Explain Gas Chromatography ?

2 MARKS

Q1. Application of HPLC ?

Q2. Application of Gas chromatography ?

UNIT –V

07 Hours

Ion exchange chromatography- Introduction, classification, ion exchange resins, properties, mechanism of ion exchange process, factors affecting ion exchange, methodology and applications

Gel chromatography- Introduction, theory, instrumentation and applications

Affinity chromatography- Introduction, theory, instrumentation and applications

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10 MARKS

Q1. Define Ion Exchange Chromatography It's classification , Application and factor affecting

5 MARKS

- Q1. Explain Ion exchange resins It's properties and mechanism ?
Q2. Write a short note on Gel Chromatography

2 MARKS

- Q1. Define Affinity Chromatography .
Q2. Application of Ion - Exchange resin .
Q3. Define Gel chromatography .