

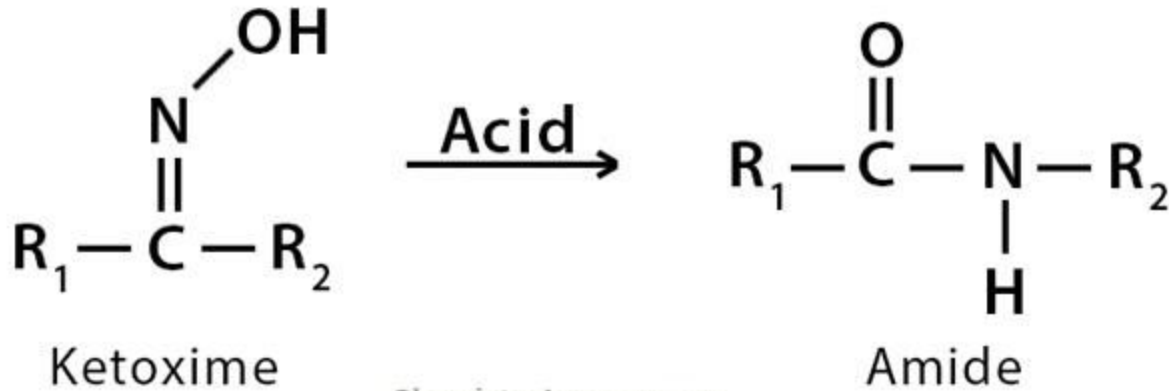
# DEPTH OF BIOLOGY

## BECKMANN REARRANGEMENT

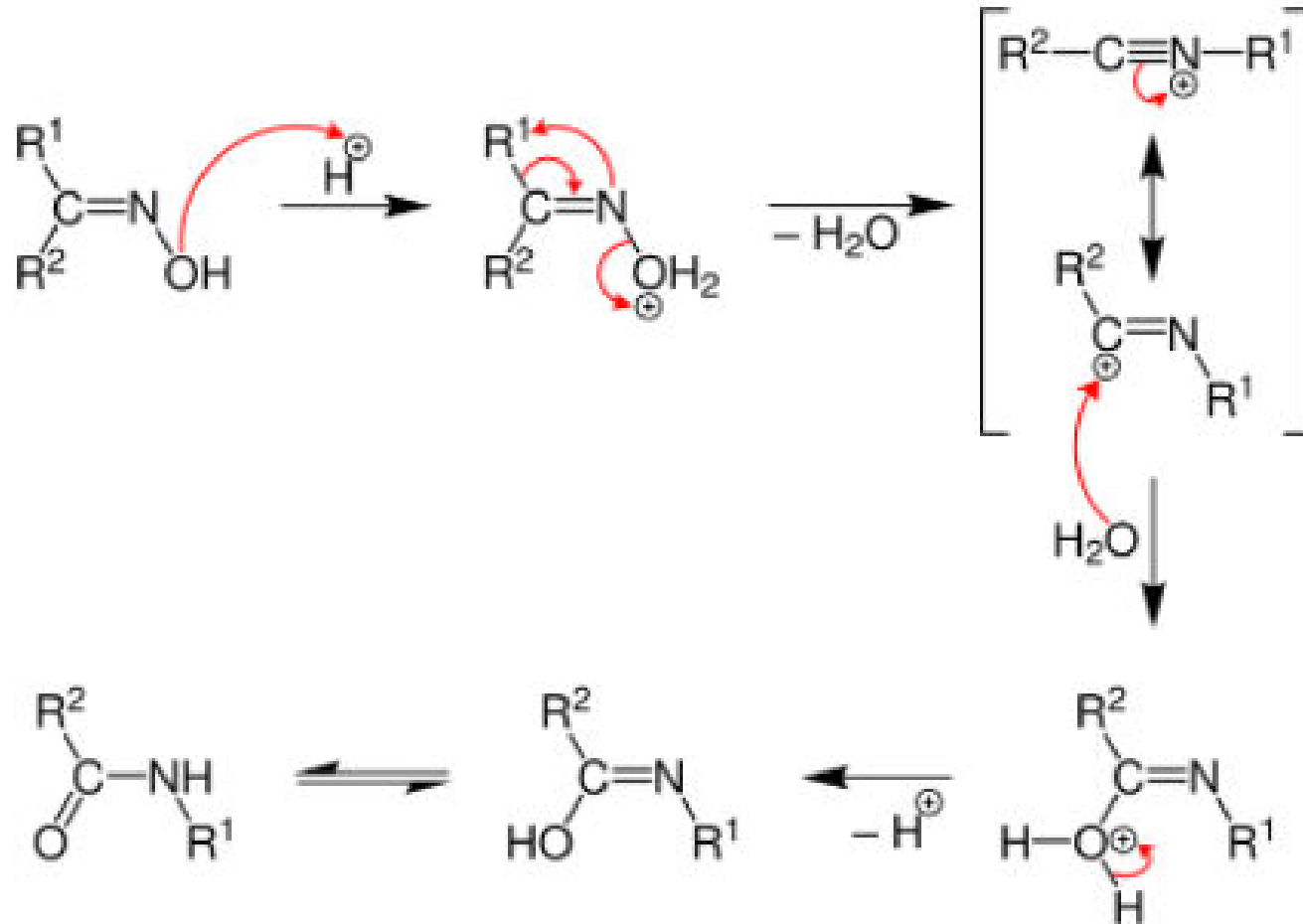
# DEPTH OF BIOLOGY

- It is a chemical process in which ketoxime is converted into secondary amide
- This takes place in the presence of concentrated  $\text{H}_2\text{SO}_4$  and  $\text{PCl}_5$
- $\text{PCl}_5$  is an electrophile species
- First the product is formed in enol form then converted into keto form to maintain stability

# DEPTH OF BIOLOGY

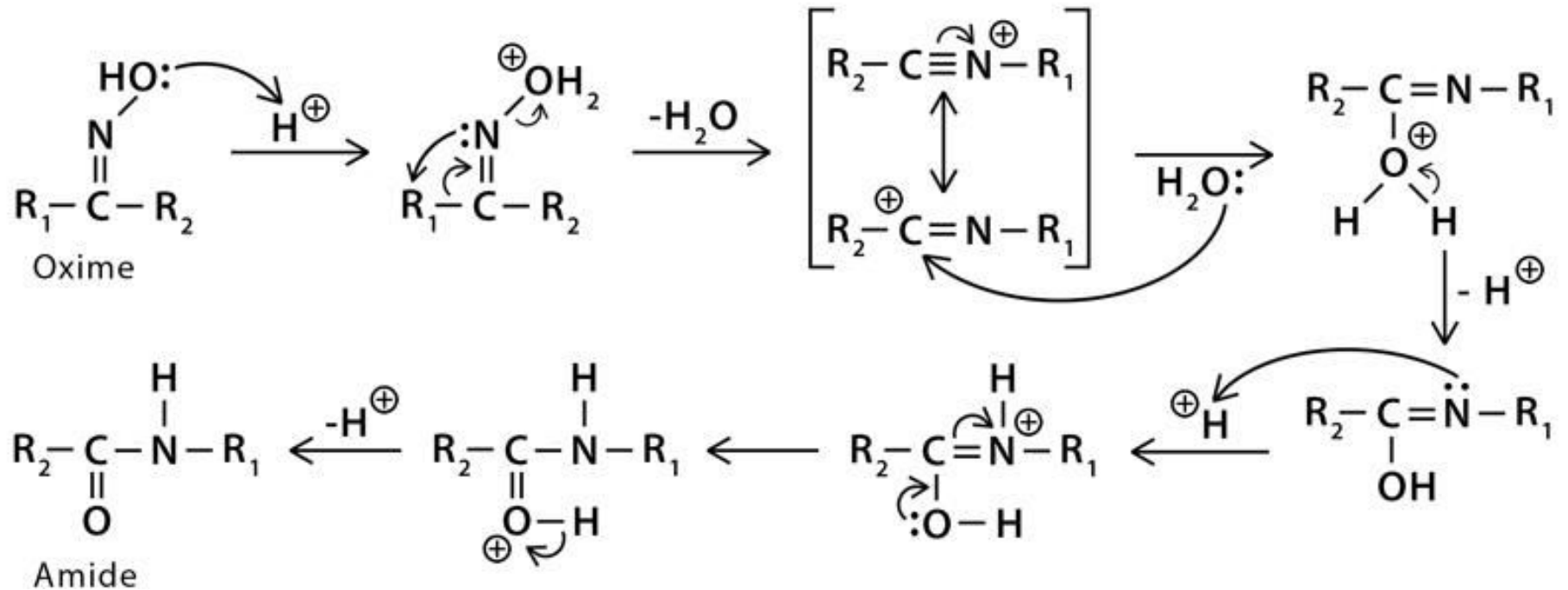


# DEPTH OF BIOLOGY



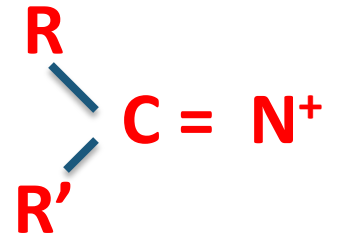
# DEPTH OF BIOLOGY

## Mechanism of Beckmann Rearrangement



# DEPTH OF BIOLOGY

- First the  $\text{OH}^-$  and  $\text{H}^+$  combine to form  $\text{H}_2\text{O}$
- $\text{H}_2\text{O}$  gets removed
- Later the compound is highly unstable as N is carrying positive charge
- So  $\text{R}'$  gets attached to  $\text{N}^+$
- this compound gets attacked by  $\text{H}_2\text{O}$
- enol form is obtained



# DEPTH OF BIOLOGY

- The enol form is converted into keto form and stable secondary amide is obtained