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GPAT QUESTION PAPER 1992 WITH ANSWER KEY

PY-PHARMACEUTICAL SCIENCES

Time: 3 hours Maximum Marks: 200

- N. B. (1) This question paper contains two parts, A and B.
 - (2) Answer all questions from Part A.
 - (3) Answer any 20 questions from Part B.
 - (4) There will be no negative marking.



- N.B. (1) There are 2 Section in this part.
 - (2) Answer all question in both Section 1 and 2.
 - (3) Answer should be given in serial order in the answer book.
 - (4) Do not skip questions while writing the answers.
 - (5) Write the question number and show your answer by writing the alphabet (Against the No.) in capital letters.
 - (6) In section 1 each question carries 1 mark
 - (7) In section 2 each question carries 2 marks.
 - (8) A model is shown at the beginning of each section in part A.
 - (9) Answer to the question in this must be written in the first 3 (three) pages of the answer books only.



Multiple choice Questions

- 1.1 Simethicone is a component of several antacid formulations. Chemically it is
 - (a) Wax
- (b) Fat
- (c) Aldehyde
- (d) Silicon

- 1.2 The pharmacy Council of India is constituted by the
 - (a) Central Government

(b) State Government

(c) Parliament

- (d) Legislative Assembly
- 1.3 All of the following physicochemical constants are useful in predicting the solubility of adrug except
 - (a) Dielectric constants

(b) pH of a solution

(c) pK_a of the drug

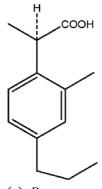
(d) Valency

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1.4	Sigma blade mixers are co	ommonly used in						
	(a) Wet granulation		(b) Dry granulation					
	(c) Powder mixing		(d) Crude fibre mixing					
1.5	The 5β pregnane is said	to have a						
	(a) Trans-anti-trans-anti-	trans backbone	(b) Cis-anti-trans-anti-tran	s backbone				
	(c) Cis-syn-trans-syn-tran	ıs backbone	(d) Trans-anti-cis-anti-cis l	packbone				
1.6	Many drugs are chiral In	a synthesis of chir	ral drug molecules in symme	etric environment				
	(a) Always one enantiom	er is obtained						
	(b) Always both enantior	ners is obtained in	equal amounts					
	(c) Always both enantion	ners is obtained in	unequal amounts					
	(d) None of the above							
1.7	Poorly manufactured tabl	ets may have smal	l pinholes on the surface. The	nis phenomenon is known as				
	(a) Picking (b)	Mottling	(c) Leaching (d) Crack	ring				
1.8	Ascorbic acid exists in na	iture						
	(a) Only in the reduced f							
	(b) Only in the oxidized form which has only biological activity							
	(c) In both reduced and activity	the oxidized form	ns in the state of reversible	equilibrium which has biological				
	(d) None of the above							
1.9	In the stable conformatio	n of 5 á -pregnane						
	(a) Rings A, B, C are in bo	oat conformation						
	(b) Rings A and B are in							
	(c) Ring A is in boat while							
	(d) All the three rings are							
1.10		parations, which o	ne will be the most irritatin					
	(a) Purified water		(b) 0.7% NaCl sol					
	(c) 0.9% NaCl solution		(d) 1% NaCl solut	ion				
1.11	In case of hypothyroidisn	n, the preferred th	yroid preparation is					
	(a) Levothyroxine		(b) Dextrothyroxi	ne				
	(c) Leothyroxine		(d) None of the ab	oove				
1.12	D-Fructose on simple red	uction gives						
	(a) L-Fructose		(b) Only Sorbitol					
	(c) Mannitol		(d) Mixture of Mannitol and Sorbitol					
1.13	Lugol's solution contains	5% of iodine. Hov	much of Lugol's solution i	s administeredto a patient thrice				
	daily to provide 60 mg of	iodine daily?						
	(a) 0.2 ml	(b) 0.3 ml	(c) 0.4 ml	(d) 0.5 ml				
1.14	The anticoagulant Hepari	n is obtained from						
	(a) Sheep's lung	(b) Dogʻs kidney	(c) Rabit's heart	(d) Rat's uterus				

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- 1.15 Which one of the following types of adverse drug reactions are not believed to be doserelated phenomenon?
 - (a) Side effects and toxic reactions
 - (b) Toxic reactions and hypersensitivity
 - (c) Side effects and hypersensitivity
 - (d) Hypersensitivity and idiosyncrasy
- 1.16 The structure of a drug having an asymmetric center is Using the IUPAC system, the configuration will be



(a) R

(b) S

(c) a

(d) β

- 1.17 Cryoscopic method is familiar in the calculations of isotonic solutions. This method isbased on
 - (a) Freezing point depression of the drug
- (b) Molecular concentration of the drug

(c) pH of the drug

- (d) None of the above
- 1.18 One thousand nanogram equal to one
 - (a) Centrigram
- (b) Gram
- (c) Kilogram
- (d) Microgram
- 1.19. Biological role of thiamine is because of facile formation of
 - (a) Thiamine hydrochloride

(b) Thiamine pyrophosphate

(c) Thiamine sulphate

- (d) None of the above
- 1.20. Infected blood products may produce serum hepatitis due to the presence of
 - (a) Hepatitis A virus

(b) Hepatitis B virus

(c) Hepatitis C virus

- (d) None of the above
- 1.21.pH of a buffer system can be calculated by using
 - (a) pH partition theory

- (b) Noyes-Whilney law
- (c) Henderson-Hasselbalch equation
- (d) None of the above
- 1.22. Osmolality measures the total number of particles dissolved in a of water and depends on the electrolytic nature of the solute.
 - (a) Kilogram
- (b) Kilolitre
- (c) Litre
- (d) Specified quantity

- 1.23. Ergot is the sclerotium of
 - (a) Fungus Claviceps purpurea

(b) Fungus Claviceps notatum

(c) Strychnos Mixpotatorm

- (d) Fungus Pencillium chrysogenum
- 1.24.A highly sensitive semiquantitative method of detecting microbial antigen in biological fluid is done by
 - (a) Radioimmuno electrophoresis
- (b) Counter immunoelectrophoresis

(c) H.P.L.C

(d) Freeze dried centrifugal method

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1.25 The glass electrode used in pH measurements is

(a) Metal-metaloxide electrode

(b) A membrane electrode

(c) Ion selective electrode

(d) None of the above

1.26 In phenonthiazine tranquillizing agents, replacement of C-2 hydrogen by chlorine

(a) Decreases activity

(b) Increases activity

(c) Activity unaffected

(d) Leads to decreased penetration into the CNS

1.27 The loading dose of a drug is based upon the

- (a) Time taken for complete elimination
- (b) Percentage of drug excreted unchanged in urine
- (c) Percentage of drug bound to plasma protein
- (d) Apparent volume of distribution and the desired drug concentration in plasma

1.28 Conformation of drugs is commonly determined by

(a) NMR

(b) NMI

(c) Mass spectrometry

(d) pH determination

1.29 Aminophylline solutions on exposure to air may develop

(a) Crystals of theophylline

- (b) Precipitate of aminophylline
- (c) Precipitate of ethylenediamine
- (d) Straw colour

1.30 The hypotensive effect of clonidine is due to its action on

(a) Beta -adrenergic receptor

(b) Alpha-adrenergic receptor

(c) H₂ -receptor

(d) H2 receptor

1.31 A step in Vitamin A synthesis is The product obtained will be



The product obtained will be

- (a) 3-hydroxy-3-methyl-1-pentene-4-yne
- (b) hex-1-yn-5-one
- (c) 3-amino-3-methyl-1-pentene-4-yne
- (d) None of the above

1.32 Sodium nitroprusside is one of the most potent blood-pressure lowering drugs. Its useis limited because of

(a) Its short duration of action

(b) Very long duration of action

(c) Ineffective of oral route

(d) None of the above

1.33 Cocaine is a monoacid tertiary base which on treatment with hot dilute acids gives

- (a) Ecogonine, methyl alcohol and scopic acid
- (b) Ecogonine, methyl alcohol and cinnamic acid
- (c) Ecogonine, methyl alcohol and benzoic acid
- (d) Ecogonine, ethyl alcohol and benzoic acid

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1.34 Use of Isoniazid is restricted due to

- (a) Ototoxicity
- (c) Neutrotoxicity

- (b) Hepatotoxicity
- (d) Bone marrow depression

1.35 Diosgenin is

- (a) An alkaloid obtained from dioscorea
- (b) A carbohydrate obtained from disscorea
- (c) A glycoside obtained from dioscorea
- (d) None of the above

1.36 The IUPAC nomenclature of the sulindac analogue

- (a) (Z)-5-Fluoro-2-methyl-1-phenylmethylene-1H-indene-3-acetic acid
- (b) (E)-5-Fluoro-2-methyl-1-phenylmethylene-1H-indene-3-acetic acid
- (c) 5-Fluoro-2-methyl-1-phenylmethylene-1H-indene-3-acetic acid
- (d) (R)-5-Fluoro-2-methyl-1-phenylmethylene-1H-indene-3-acetic acid

1.37 Bubble point test is done to determine

- (a) The surface tension of the liquid in capillary tubes
- (b) The viscosity of the liquid in ampoules http://www.xamstudy.com
- (c) The pH of a 1% solution
- (d) The volume of the solution stored in a specified container

1.38 The "Hemiacetal" form of aldosterone is between

- (a) C-11, β -hydroxyl and C-20 carbonyl
- (b) C-11, β -hydroxyl and C-21 hydroxy
- (c) C-11, β -hydroxy and C-18 carbonyl
- (d) C-21, hydroxyl and C-20 carbonyl

1.39 Surfactants are characterized by the presence of

- (a) Water solubilizing and fat solubilizing groups in the same molecule
- (b) Only negative charges
- (c) Only positive charges
- (d) None of the above

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- 1.40 Acetous perchloric acid solution a common titrant in non-aqueous titrimetry is standardized by
 - (a) Standard alcoholic KOH solution
 - (b) N/10 Potassium permanganate
 - (c) Potassium hydrogen phthalate solution in glacial acetic acid
 - (d) Mercuric acetate in glacial acetic acid



- 2.1 The antibiotics mentioned below are obtained from the organisms listed from A to E. match them.
 - (1) Neomycin
 - (2) Gentamycin
 - (3) Bacitracin
 - (4) Tobramycin
 - (a) 1-A, 2-C, 3-B, 4-D
 - (c) 1-A, 2-B, 3-D, 4-C

- (A) Streptomyces fradiae
- (B) Micromonospora purpurea
- (C) Streptomyces tenebrarius
- (D) Bacillus subtilis
- (E) Bacillus polymyxa
- (b) 1-B, 2-D, 3-C, 4-A
- (d) 1-B, 2-E, 3-C, 4-A
- 2.2. Given below is a list of medicinal plants. Match them correctly with the list of constituents given in A to E.
 - (1) Holarrhenaa ntidysenterica
 - (2) Cymbopogan flexuous
 - (3) Urginea indica
 - (4) Linum usitatissiumum
 - (a) 1-A, 2-C, 3-B, 4-D
 - (c) 1-A, 2-B, 3-D, 4-C

- (A) Conessine
- (B) Citral
- (C) Mucilage
- (D) Cocaine
- (E) Scillarenin
- (b) 1-A, 2-B, 3-E, 4-C
- (d) 1-B, 2-E, 3-C, 4-A
- 2.3. The following drugs are included under the schedules listed in A to E. match them.
 - (1) Meprobamate

(A) Schedule E (B) Schedule FF

- (2) Poisonous drugs
- (3) Ophthalmic preparations
- (4) Biological and special products
- (C) Schedule C and C
 - (D) Schedule X
 - (E) Schedule Q

(a) 1-D, 2-A, 3-B, 4-C

(b) 1-B, 2-D, 3-C, 4-A

(c) 1-A, 2-B, 3-D, 4-C

(d) 1-B, 2-E, 3-C, 4-A

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2.4.	Foll	owing drugs contain 2 hydroxyl groups	each	. Nature of these hydroxyl groups are indicated in A to E.		
	Mat	tch them correctly.				
	(1)	Morphine	(A)	Akoholic but one 1 another 2		
	(2)	Chloramphenicol	(B)	Akoholic and both 1		
	(3)	Apomorphine	(C)	Both phenolic		
	(4)	Cortisone	(D)	One alcoholic and one phenolic		
			(E)	Akoholic but 1 and another 3		
	(a)	1-A, 2-C, 3-B, 4-D	(b)	1-D, 2-A, 3-C, 4-E		
	(c)	1-A, 2-B, 3-D, 4-C	(d)	1-B, 2-E, 3-C, 4-A		
2.5.	Γhe	following drug molecules contain hetero	ocycli	c rings listed in A to E. match them correctly.		
	(1)	Haloperidol	(A)	Pyrimidine		
	(2)	Sulphadiazine	(B)	Pyridine		
	(3)	Amiloride	(C)	Piperidine		
	(4)	Pheniramine	(D)	Pyrazine		
			(E)	Pyridazine		
	(a)	1-C, 2-A, 3-D, 4-B	(b)	1-B, 2-D, 3-C, 4-A		
	(c)	1-A, 2-B, 3-D, 4-C	(d)	1-B, 2-E, 3-C, 4-A		
2.6. Following drugs exhibit their action by enzyme inhibition. Enzymes are listed in A to E. Match them correct						
	(1)	Captopril	(A)	β – lactamase		
	(2)	Clavulanic acid	(B)	MAO		
	(3)	Pargyline	(C)	Monooxygenase		
	(4)	Methozolamide	(D)	Carbonic anhydrase		
			(E)	ACE		
	(a)	1-C, 2-A, 3-D, 4-B	(b)	1-E, 2-A, 3-B, 4-D		
	(c)	1-A, 2-B, 3-D, 4-C	(d)	1-B, 2-E, 3-C, 4-A		
2.7.	Foll	owing preparations are assayed by bio	logica	d techniques using the animal or its parts listed in A to E.		
	mat	tch them correctly.				
	(1)	Cod liver oil	(A)	Sheep blood		
	(2)	Heparin injection	(B)	Rabbit		
	(3)	Oxytocin injection	(C)	Rat		
	(4)	Insulin injection	(D)	Anaesthetized chicken		
			(E)	Cat		
	(a)	1-D, 2-A, 3-C, 4-B	(b)	1-B, 2-D, 3-C, 4-A		
	(c)	1-A, 2-B, 3-D, 4-C	(d)	1-B, 2-E, 3-C, 4-A		
2.8. F	ollo	wing I.P. assays involve the principles li	sted i	n A to E. Match them.		
		Sodium chloride injection		Titration with N/10 iodine		
(2) ′	Trimethoprim	(B)	Oxidation involving 2 : 6 dichlorophenol indophenol		

(C) Argentometry

(D) Non-aqueous(E) Acidimetry

(3) Analgin tablets

(4) Ascorbic acid

(a)	1-C, 2-A, 3-D	(b)	1-B, 2-D, 3-C						
	1-A, 2-B, 3-D	(d)	1-B, 2-E, 3-C						
2.9. Giv	en below are some antihypertensive m	ech	anisms. Drugs which are closely associated with these						
me	mechanisms of action are listed in A to E. Match them correctly?								
(1)	Ganglion blocking	(A)	Methyl dopa						
(2)	Catecholamine depletor	(B)	Hydralazine						
(3)	False neurotransmitter	(C)	Reserpine						
(4)	Direct action on arterioles	(D)	Mecamylamine						
		(E)	Veratrum alkaloids						
, ,	1-C, 2-A, 3-D, 4-B	(b)	1-B, 2-D, 3-C, 4-A						
-	1-A, 2-B, 3-D, 4-C		1-D, 2-C, 3-A, 4-B						
2.10.List	ed are Vitamins. Their associations with	cert	ain coenzymes are well known. The names of coenzymes						
are	given in A to E. match them correctly.								
(1)	Thiamine	(A)	Co-carboxylase						
(2)	Riboflavin		Co-enzyme A						
(3)	Panothenic acid		NAD						
(4)	Nicotinamide		FAD						
			ATP						
	1-A, 2-D, 3-B, 4-C		1-B, 2-D, 3-C, 4-A						
	1-A, 2-B, 3-D, 4-C		1-B, 2-E, 3-C, 4-A						
2.11.Listed are some of the crude drugs which are tested for the active constituents by the tests mentioned in									
	E. Match them correctly.								
	Cinchona Bark		Fluorescene test						
	Nux vomica seeds		Keller Killiani						
(3)	Digitalis leaves	(C)							
(4)	Senna leaves		Mayer's test						
			Sham's test						
	1-A, 2-D, 3-B, 4-C		1-B, 2-D, 3-C, 4-A						
(c)	1-A, 2-B, 3-D, 4-C	(d)	1-B, 2-E, 3-C, 4-A						
2.12.List	ted are some of the common volatile oil	s. Th	neir active constituents are given in A to E. Match them						
cor	rectly.								
(1)	Peppermint oil	(A)	(+)-Limonene						
(2)	Turpentine oil	(B)	1:8-Cineole						
(3)	Eucalyptus oil		α Pinene						
(4)	Lemon oil		(-) Menthol						
			(+) Methol						
(a)	1-C, 2-A, 3-D, 4-B		1-D, 2-C, 3-B, 4-A						
(c)	1-A, 2-B, 3-D, 4-C	(d)	1-B, 2-E, 3-C, 4-A						

2.13.Ma	tch the each pair with the type of a struc	tural	relationship they exhibit.
(1)	(R) and (S) Naproxen	(A)	Tautomers of one another
(2)	Dilactim and Monolactim	(B)	Diastereomers of one another form of Barbituric acid
(3)	Quinine and Quinidine	(C)	Non-superimposable mirror images of each other
(4)	Eclipsed and staggered form	(D)	Superimposable mirror images of each other of
			phenothiazine about side chain carbon oecarbon bond
		(E)	Conformational isomers of one another
(a)	1-C, 2-A, 3-B, 4-E	(b)	1-B, 2-D, 3-C, 4-A
(c)	1-A, 2-B, 3-D, 4-C	(d)	1-B, 2-E, 3-C, 4-A
2.14.Vai	rious polymers used in pharmacy are giv	en. N	Match with the respective monomers A to E.
(1)	Carbopol	(A)	Methacrylate ester
(2)	Eudragits	(B)	Ethylene
(3)	Polyethylene	(C)	Ethylene glycol
(4)	Polycarbonate	(D)	(Bis-phenol + phosgene)
		(E)	Acrylic acid
(a)	1-C, 2-A, 3-D, 4-B	(b)	1-B, 2-D, 3-C, 4-A
(c)	1-A, 2-B, 3-D, 4-C	(d)	1-E, 2-A, 3-B, 4-D
2.15.Fol	lowing are some naturally occurring sub	ostan	ces. They are classified under different categories which
are	e listed in A to E. match them correctly.		
(1)	Prostaglandins	(A)	Opioids
(2)	Codeine	(B)	Eicosinoids
(3)	Angiotensin II	(C)	Corticoids
(4)	Strophanthidin	(D)	Peptide
		(E)	Cardinolide
(a)	1-C, 2-A, 3-D, 4-B	(b)	1-B, 2-A, 3-D, 4-E
(c)	1-A, 2-B, 3-D, 4-C	(d)	1-B, 2-E, 3-C, 4-A
2.16.Fol	llowing are some of the analytical instru	ment	s. Their important components are listed in A to E. Match
the	em correctly.		
(1)	HPLC	(A)	Monochromator
(2)	IR double beam spectrophotometer	(B)	Dropping mercury electrode
(3)	Karl-Fischer titrator	(C)	Isocratic pump
(4)	Polarograph	(D)	Platinum electrode
		(E)	Polariser
(a)	1-C, 2-A, 3-D, 4-B	(b)	1-B, 2-D, 3-C, 4-A
(c)	1-C, 2-A, 3-D, 4-B	(d)	1-B, 2-E, 3-C, 4-A
2.17.The	e hard gelatin capsule sizes are mention	ed in	their number. Their approximate capacity are listed in
A to	o E. Match their correct volume.		
(1)	0	(A)	0.10 ml
(2)	1	(B)	0.15 ml
(3)	3	(C)	0.30 ml

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(4) 5

(D) 0.55 ml

(a) 1-C, 2-A, 3-D, 4-B

(E) 0.75 ml

(-) - -, - -, - -

(b) 1-B, 2-D, 3-C, 4-A

(c) 1-A, 2-B, 3-D, 4-C

- (d) 1-E, 2-D, 3-C, 4-B
- 2.18.Listed are some of the commonly used drugs. Their pharmacological actions are listed in A to E. match
 - (1) Aspirin

(A) Rises body temperature

(2) Acetaminophen

(B) Non-analgesic anti-inflammatory

(3) Phenylbutazone

(C) Non-anti-inflammatory analgesic

(4) Probenacid

(D) Increases depth of respiration(E) Increases fluid retention

(a) 1-C, 2-A, 3-D, 4-B

(b) 1-B, 2-D, 3-C, 4-A

(c) 1-A, 2-B, 3-D, 4-C

- (d) 1-D, 2-C, 3-B, 4-E
- (C) 1-A, 2-B, 3-D, 4-C
- 2.19. Match the preservations/antioxidants listed in A to E for the preparations mentioned below.
 - (1) Anaesthetic ether

(A) Chloroform

(2) Formaldehyde

- (B) Ethylenediamine
- (3) Injection of Adrenaline
- (C) Paraformaldehyde
- (4) Injection of Aminophylline
- (D) Sodium bisulphate

(a) 1-C, 2-A, 3-D, 4-B

(E) Alcohol

(4) 1 4, 2 1, 4 2, 1 2

(b) 1-B, 2-D, 3-C, 4-A

(c) 1-E, 2-C, 3-D, 4-B

- (d) 1-B, 2-E, 3-C, 4-A
- $2.20. In \ communition, \ certain \ type \ of \ materials \ listed \ in \ A \ to \ E \ are \ not \ suitable \ for \ the \ mills mentioned \ below.$

Match them.

(1) Cutter mill

(A) Soft material

(2) Hammer mill

(B) Adhesive material

(3) Revolving mill

(C) Friable material

(4) Fluid energy mill

(D) Liquifiable material

(a) 1-C, 2-A, 3-D, 4-B

(E) Abrasive material

(c) 1-E, 2-C, 3-D, 4-B

- (b) 1-B, 2-D, 3-C, 4-A (d) 1-B, 2-E, 3-C, 4-A
- PART B

- 3. Define:
 - (A) Co-solvency
- (B) Hydrotrophy
- (C) Eutectic mixtures
- 4. Complete the following reactions giving the structures

$$\alpha$$
-Pinene $\xrightarrow{\text{HCl (gas)}}$ A $\xrightarrow{\text{CH}_3\text{COONa}}$ B $\xrightarrow{\text{HCOOH}}$ C \longrightarrow D $\xrightarrow{\text{O}_2/\text{Ni}}$ E

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- 5. (A) Thiamin when treated with sodium sulphite solution saturated with SO_2 at room temperature, decomposes quantitatively into 2 components. What are they? Give their structural formulae.
 - (B) Riboflavin on exposure to light in sodium hydroxide solution forms an insoluble product. What is the product? Write equation. Why is it insoluble?

(C) Caffeine
$$\xrightarrow{\text{Cl}_2}$$
 A $\xrightarrow{\text{CH}_3\text{OH}}$ B $\xrightarrow{\text{DiHCl}}$ C + CH₃Cl $\xrightarrow{\text{CH}_3\text{I}}$ D $\xrightarrow{\text{NaOH}}$ D

6. (A) Name two common equipments used for testing the hardness of a tablet

(i)

(ii)

- (B) Give four factors which affect the hardness of a tablet
- (C) Why friability test is performed? How is it performed?
- 7. Give four important tests to detect the emulsion types.
- 8. (A) List the names of three important semisynthetic hydrocolloids used in pharmacy.
 - (B) What is their chemical nature?
 - (C) Give three important uses of the above hydrocolloids.
- 9. (A) Diethyl malonoate is treated with excess of n-propybromide in presence of sodium ethoxide to give intermediate (A). The intermediate (A) on refluxing in dilute alkali gives anticonvusant drug. Write the reactions and structures.
 - (B) 2-Aminopyridine is reductively alkylated using 1 mol of p-methoxybenzaklehyde and reducing agent to give an intermdieate(B). the intermediate (B) on treatment with 1 mol dimethylaminoethylbromide in presence of sodium amide gives an antihistaminic drug. Write the reactions and structures.
 - (C) Arrange the nitrogens in the drug referred in
 - (d) Above in decreasing order of basicity.
- 10. (A) Give the graph [include correct scale and values] of cardiac action potential as recorded from a Purkinje fibre. Indicate the phases of depolarization and repolarisation (graph paper not to be used).
 - (B) Expand the abbreviations and indicate how they are formed. http://www.xamstudy.com
 - (i) cAMP
 - (ii) GABA
- 11. Write the appropriate reagent(s) and the structures in the following transformations. Give the trivial name of the drug and the category it belongs.

(i)
$$p-Methoxyphenyldydrazine \xrightarrow{ZnCl_2} A + NH_3 + H_2O$$

Т.

Ethyl 4-oxopentanoate

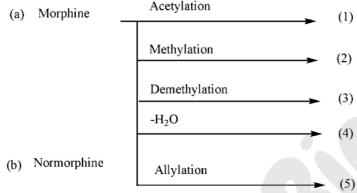
(ii) A + p - chlorobenzoylchloride $\xrightarrow{\text{Base}} B$

(iii) $B \longrightarrow [Drug]$ (contains – COOH)

Note: One step involves Fischer indolisation.

12.	(A) Give synthesis of pyridoxine starting from 5-ethoxy-4-methyl isoxazole and cis butenediol protected
	with isobutaraldehyde.

- (B) Name the key reaction
- 13. Give the names of products.



- (c) Give the structural formula for Nalorphine
- 14. (A) What is Vitali's test? Which natural products are distinguished by this test?
 - (B) Give reasons for the following:
 - (i) Morphine is soluble in aqueous sodium hydroxide.
 - (ii) In the assay of alkaloids, the final residue is treated with little alcohol before it is finally dissolved in acid and titrated.
- 15. (A) Alkaloids of ergot exist in stereoisomeric pairs. Which are they? Whine one is pharmacologically active?

 - (D) *Catharanthus roseus* contains 2 alkaloids which are effective in the treatment of neoplastic Diseases, they are
 - (7)(8)
- 16. (A) Give the procedure for IP assay for Nikethamide injection.
 - (B) Give the reactions involved in the above assay.
 - (C) To which category of drugs this compound belongs?
- 17. (A) Show the 5β -cis fusion and 5α -trans fusion in case of steroid nucleus. What they are usually called?
 - (B) Name the different types of insulin preparations.
 - (C) Usual route of administration of insulin preparation is _____
 - (D) Since insulin is a _____ it cannot be given orally.

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- 18. (A) Which are the substances defined as Narcotic drugs and Psychotropic substances under Narcotic drugs and Psychotropic Substances Act, 1985 and rules?
 - (B) What is the international name for small-pox vaccine?
 - (C) How much of sodium chloride is required to render 150 ml of a 2% solution of procaine hydrochloride isotonic with blood serum? [Freezing point depression of 1% procaine hydrochloride is - 0.12°C and 1% sodium chloride is - 0.56°C].
- 19. What are the particulars to be recorded in analytical records for parenteral preparations as per the G.M.P. requirement under and C rules? List them correctly.
- 20. (a) An antibiotic isolated from streptomyces venezuelae having an aromatic nitro group, on hydrolysis gives
 - (A) Dichloroacetic acid
 - (B) 2-amino-1-p-nitrophenyl-1: 3-propanediol
 - (1) What is the structural formula of the antibiotic?
 - (2) How many asymmetric carbon atoms are present in B?
 - (3) What is the optical activity and configuration of the active from of the antibiotic?
 - (b) What are the important hydrolytic products of streptomycin? Give their names only.
- 21. (A) Explain what is E_{1cm}^{1%}
 - (B) What is Group frequency region and finger print region?
 - (C) What is retention volume?
- 22. (A) What processes are to be adopted in pharmaceutical industry for the
 - (i) separation of bacteria from the parenteral liquids
 - (ii) purification of colloids and enzymes
 - (iii) removal of particulate matter in the air?
 - (B) How much of 90% and 20% alcohols are required to produce 350 ml of 60% alcohol?
- 23. (A) Classify the surfactants with suitable example for each:
 - (B) LAL test is used as in-process control in parenteral preparations:
 - (i) What for is it used?
- (ii) How is the test performed?
- 24. How will you rectify the following defects in tablet manufacturing?
 - (i) Punch variation

- (ii) Hardness variation
- (iii) Double impression
- (iv) Poor flow
- 25. (A) Name four foam systems used in aerosol technology.
 - (B) Name four parameters in the evaluation of foam stability.
 - (C) Explain the following words used in communition in one sentence for each
 - (i) Open circuit milling
 - (ii) Closed circuit milling

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- 26. (A) In the evaluation of soft capsules, the following terms are used. Explain them in one sentence for each.
 - (i) Soft spot
- (ii) Bloating
- (iii) Foreign capsule
- (B) A drug is used for synthesis purpose in the concentration of 8 mg/kg and it is available as 2 ml ampoules of 150 mg strength. how much of the drug is required for an adult male aged 32 years having a weight 45 kg with the body surface of 1.9 m2?
- 27. (A) Synthesis of a steroid hormone is given below. Write all the structures. Give the reagents used in Oppenaur oxidation:
 - (i) 3 β -Acetoxypregna-5, 16-diene-20one $\xrightarrow{H_2(Pd)}$ A
 - $\text{(ii)} \hspace{0.2cm} \text{(A)} \xrightarrow{\hspace{0.2cm} \text{NaOH} \hspace{0.2cm} } \text{B} \hspace{0.2cm} \xrightarrow{\hspace{0.2cm} \text{Oppenaur} \hspace{0.2cm} } \text{[Hormone]}$
 - (B) The above hormone shows two absorption bands in carbonyl region in its infrared spectrum. Write the approximate position of the band in wave numbers and indicate the corresponding chromophore.

End of paper

ANSWER KEY GATE 1992

Section - A

1.1	d	1.11	b	1.21	с	1.31	a
1.2	a	1.12	d	1.22	a	1.32	a
1.3	d	1.13	С	1.23	a	1.33	С
1.4	С	1.14	a	1.24	b	1.34	С
1.5	b	1.15	d	1.25	с	1.35	С
1.6	b	1.16	a	1.26	b	1.36	b
1.7	a	1.17	a	1.27	d	1.37	a
1.8	С	1.18	d	1.28	a	1.38	С
1.9	d	1.19	b	1.29	a	1.39	a
1.10	d	1.20	b	1.30	b	1.40	a

Section - B

2.1	С	2.6	b	2.11	a	2.16	С
2.2	b	2.7	d	2.12	b	2.17	d
2.3	a	2.8	a	2.13	a	2.18	d
2.4	b	2.9	d	2.14	d	2.19	С
2.5	a	2.10	a	2.15	b	2.20	С