

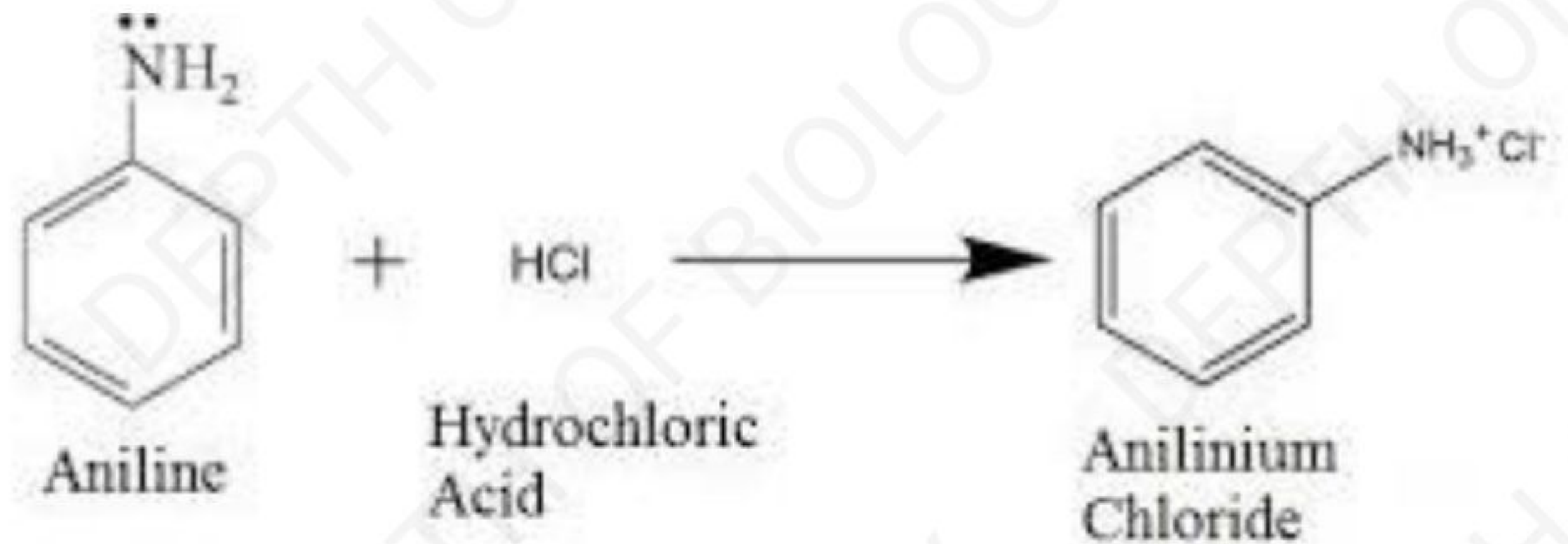
QUALITATIVE TEST OF AMINES

[DEPTH OF BIOLOGY]

Presence of amine can be confirmed by performing following tests- [DEPTH OF BIOLOGY]

(a) Solubility Test:

Amines are basic in nature, so they dissolve in mineral acids like hydrochloric acid.



[DEPTH OF BIOLOGY]

(b) Litmus Test:

Amines are basic in nature. So it turns red litmus paper blue.

(c) Carbylamine Test: [DEPTH OF BIOLOGY]

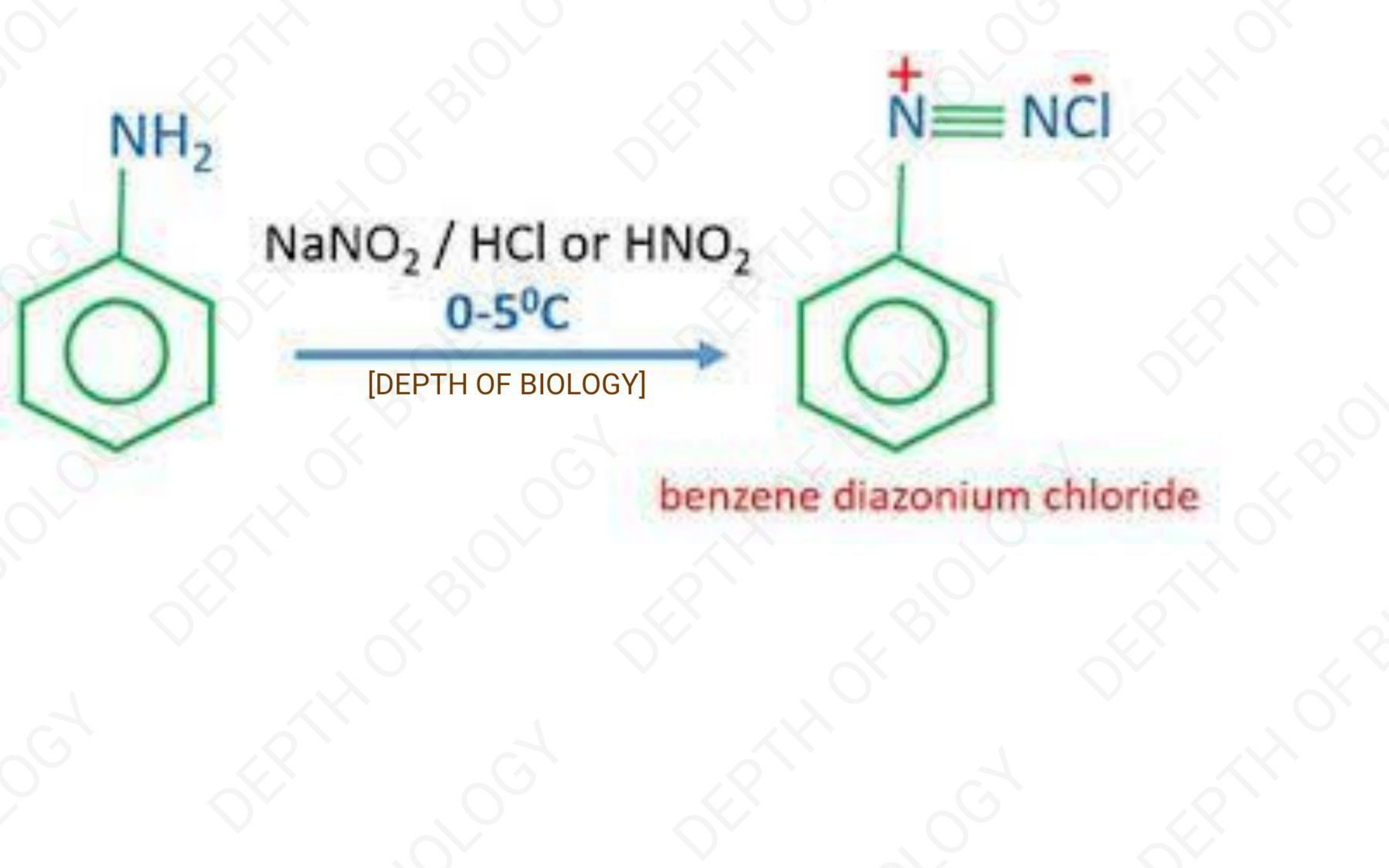
On treating primary amine with alcoholic KOH & chloroform, isocyanide (which has unpleasant odour) is produced



(d) Azo-Dye Test: [DEPTH OF BIOLOGY]

Aromatic primary amine + nitrous acid (HNO_2) produces Diazonium salt which couples with β -naphthol to form an orange coloured azo dye

[DEPTH OF BIOLOGY]

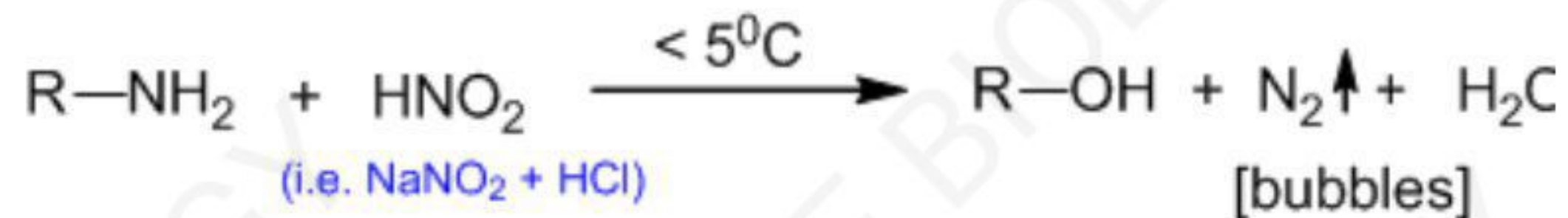


Distinguishing Test for Primary, Secondary and Tertiary Amines

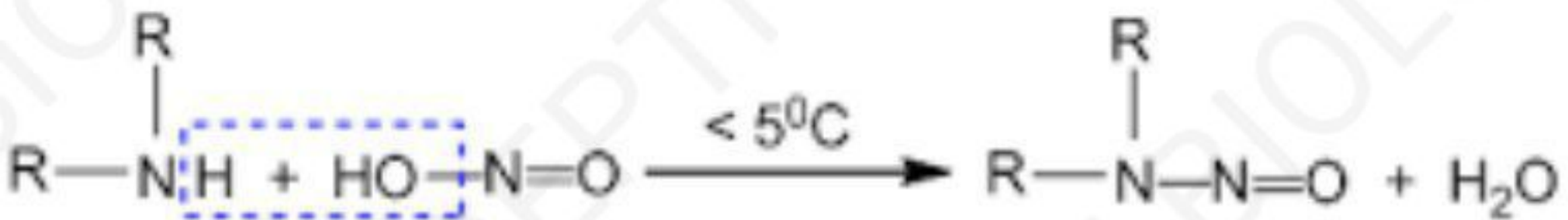
[DEPTH OF BIOLOGY]

1. NITROUS ACID TEST-

Primary amine



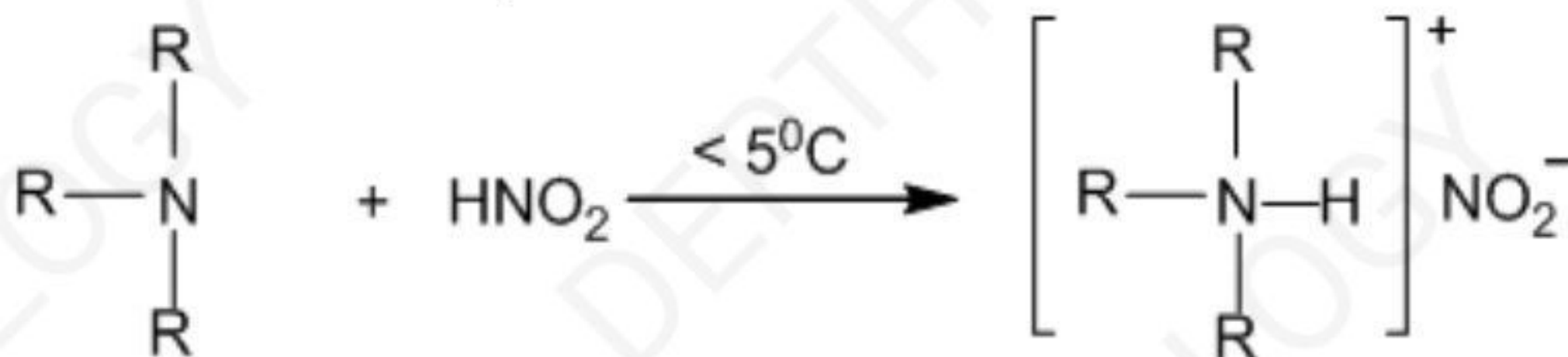
Secondary amine



**NITROSOAMINE
(YELLOW COLOURED
OILY COMPOUND)**

[DEPTH OF BIOLOGY]

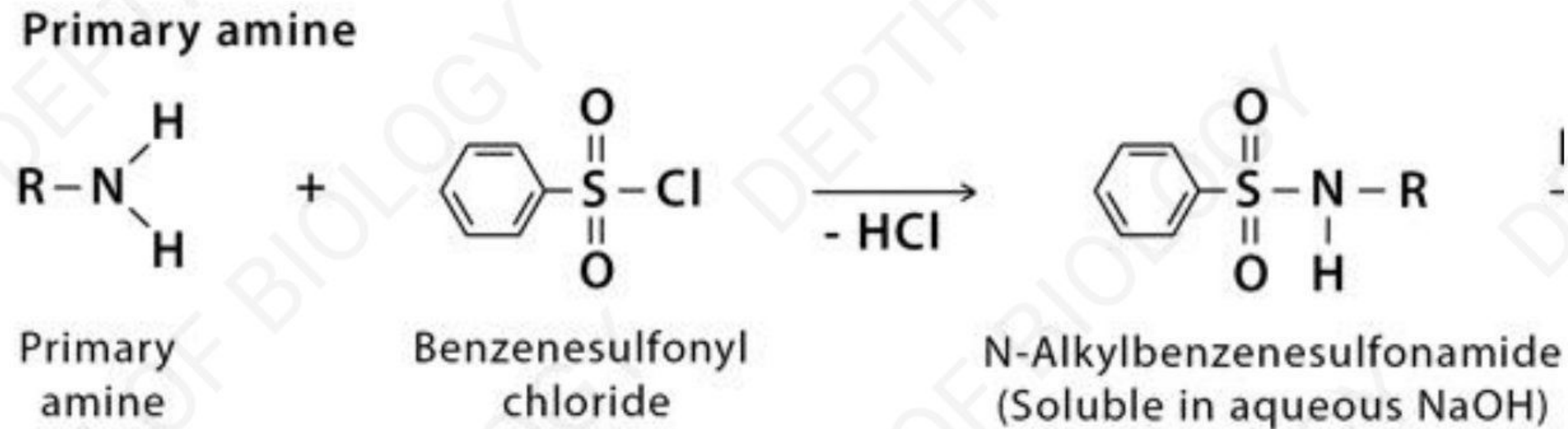
Tertiary amine



**TRIALKYL AMMONIUM
NITRATE (SOLUBLE
NITRILE SALT)**

2. HINSBERG TEST:

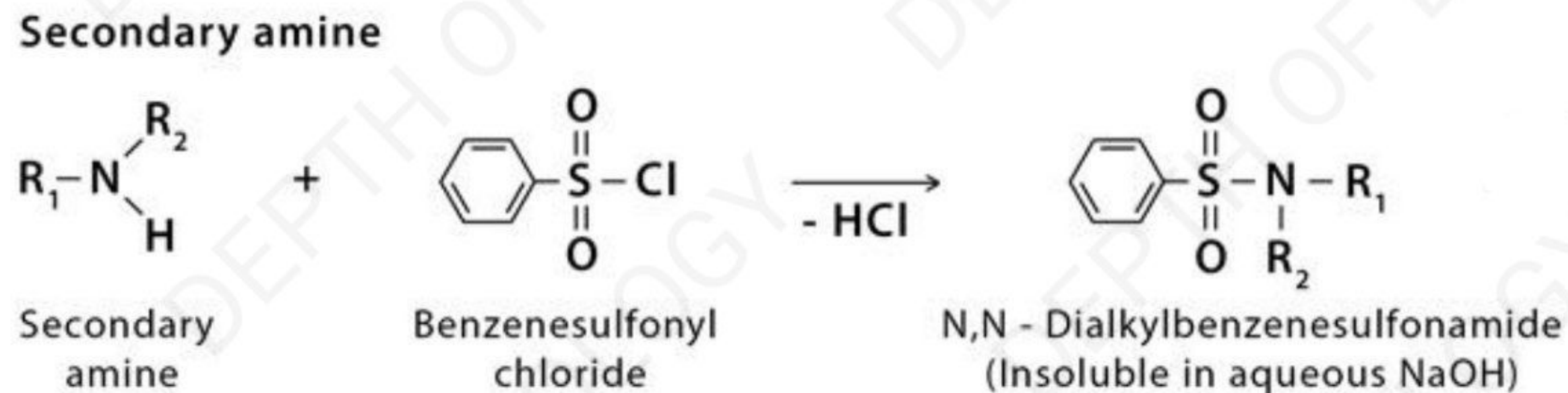
Primary amines + benzene sulfonyl chloride $\xrightarrow{-\text{HCl}}$ N-alkyl benzene sulphomamide



[DEPTH OF BIOLOGY]

[DEPTH OF BIOLOGY]

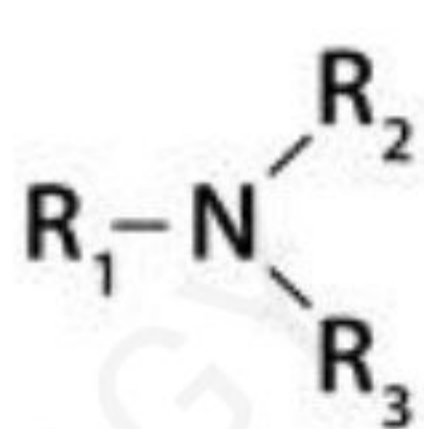
benzene sulfonyl chloride + secondary amines



[DEPTH OF BIOLOGY]

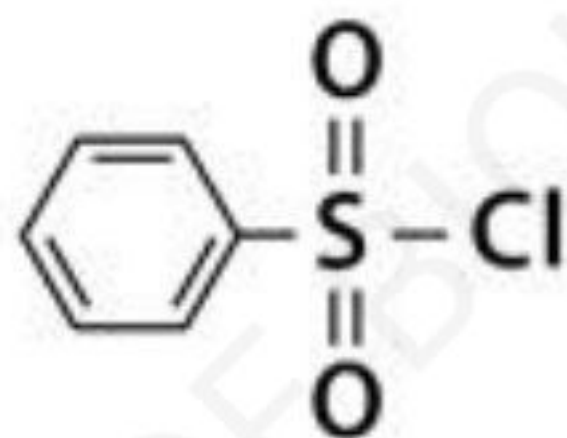
benzene sulfonyl chloride + tertiary amine= no reaction

Tertiary amine



Tertiary
amine

+



Benzenesulfonyl
chloride

[DEPTH OF BIOLOGY]



NO REACTION