

Incompatibility

→ Two or more Ingredient mix

↳ Then undesirable change

[DEPTH OF BIOLOGY]

- ↳ Physical
- ↳ Chemical
- ↳ Therapeutic

* Not occur only during compound and dispense but also at any stage :-

- (a) During formulation
- (b) Manufacturing
- (c) Packaging
- (d) Addition of drug

* Important to determine Incompatibility :-

- (1) Safety of Medicament
- (2) Efficacy
- (3) Appearance
- (4) Purpose

[DEPTH OF BIOLOGY]

Types of Incompatibility

Physical Incomp.	chemical Incomp.	Therapeutic Incomp.
* 2 or >2 substance combined unlike together * Physical changes takes place unacceptable product is formed. → Colour change → Odour → Taste → viscosity * They are usually visible and easily correctable.	* Chemical Interaction b/w Ingredient of Prescription ↓ toxic and Inactive product may formed. * chemical incomp. occur due to : → oxidation reduction → Acid base hydrolysis → combination rxn * noticed by Precipitation, effervescence, decomposition, Colour change	→ Pharmacokinetic ↓ ADME → Pharmacodynamic ↓ → Synergism → Additive → Antagonism

[DEPTH OF BIOLOGY]

Physical Incompatibility

Correction → Change the order or mixing of the prescription.
→ Emulsification
→ Adding of suspending agent.

Example → (a) Immiscibility (b) Insolubility
(c) Precipitation (d) Liquefaction

(a) Oil and H_2O → Miscible with the help of emulsifying agent.

Example → Castor oil → 15 ml [By addition of
water → 60 ml Emulsifying agent]

(b) Insoluble → Inability of material to dissolve [Many Inorg. and org. compound insoluble in particular solvent]

Example → Epid. Sulphate	—	0.25 gm
Menthol	—	0.02 ml
Liq. Paraffin	—	30 ml

[DEPTH OF BIOLOGY]

(c) Precipitation → If the solvent is added which is insoluble.

Example → Resin Insoluble in H_2O , If we add resin then agglomerate forming in diffusible precipitates.

So to prevent this, slowly add dilute tincture with fast stirring or by adding thickening agent.

(d) Liquefaction → If two solid is added whose have low melting point then soft Mass is formed.
[Eutectic mixture]

* Shown by Camphor ~~Menthol~~, Menthol, Phenol [DEPTH OF BIOLOGY]

* These type of substance create problem when they are dispersed ~~form~~ in powder form.

Example :-

Menthol — 5 gm

Camphor — 5 gm

Ammonium chloride — 30 gm

light Mg Carbonate — 60 gm

[DEPTH OF BIOLOGY]

Chemical Incompatibility

Tolerated

Adjusted

Chemical interaction can be minimised by changing the order of mixing.



No alteration is made in the formulation.

Addition or substitution of one of the reacting ingredient of a prescription with another of equal therapeutic value.

eg → Caffeine citrate

[DEPTH OF BIOLOGY]

Chemical Incompatibility :-

(a) May be Intentional

(b) UnIntentional

Example ⇒ (1) Precipitating yielding Interaction

Diffusible

For dispensing diffusible ~ method A

Indiffusible

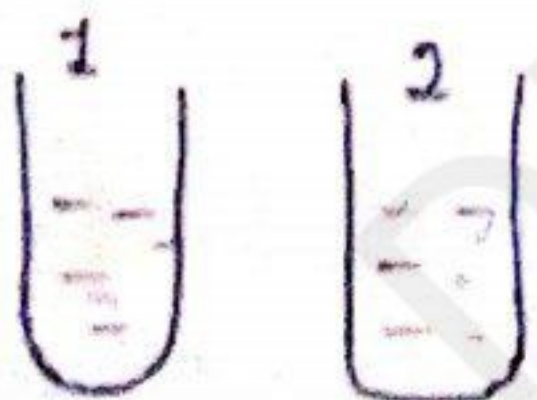


when diffusible ppt. are form in very small quantity.

For dispensing indiffusible → Method B

[DEPTH OF BIOLOGY]

Method A → The vehicle into two equal portion

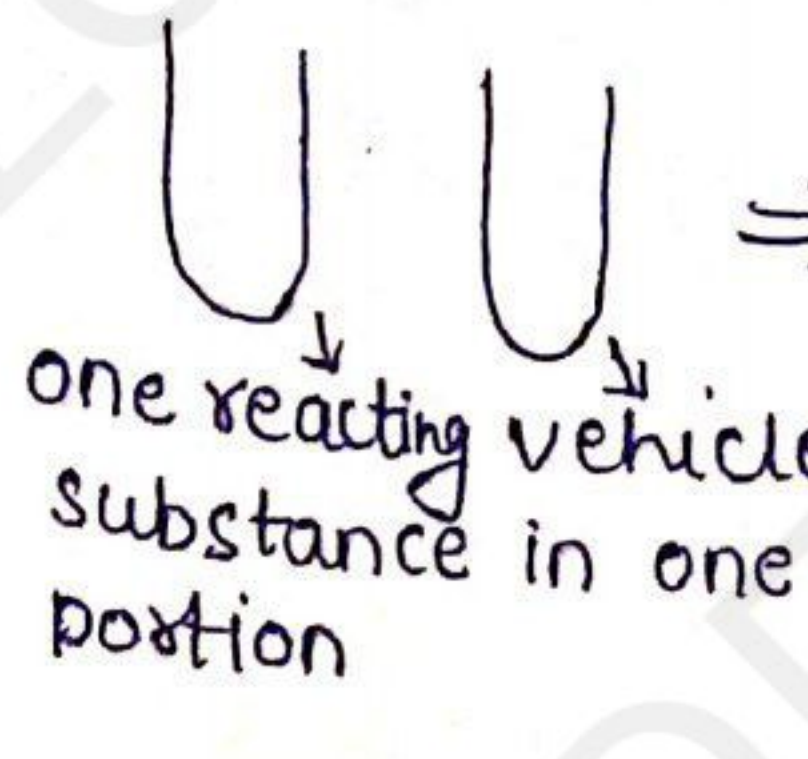


one reacting substance

other reacting substance

⇒ Mix the both by slowly adding one portion to the other with rapid stirring.

Method B $\Rightarrow$ Used when in diffusible ppt. form in long quantity.

 $\Rightarrow$ 2 g / 100 ml C.T. Powder $\rightarrow$  , Now add (mortar) second part of vehicle to produce smooth mucilage.

[DEPTH OF BIOLOGY]

$\downarrow$
Then add other reacting substance

- $\rightarrow$ Now mix 1 with 2 with rapid stirring
- $\rightarrow$ Labelling $\Rightarrow$ Shake well before use

② Soluble Iodide Incomp. $\Rightarrow$ Iodide $\xrightarrow{\text{O}}$ Iodine [undesirable]

Example $\rightarrow$ Oxidation of Iodide / ferrous iodide with Potassium chlorate.

- Potassium chlorate + Solub. Iodides $\Rightarrow$ I_2 liberate

③ Causing evolution of CO_2

[DEPTH OF BIOLOGY]

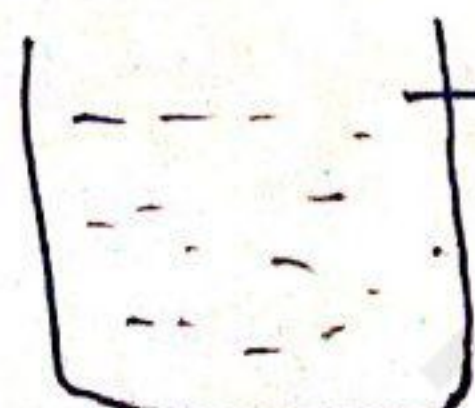
Eg $\rightarrow$ Bismuth Sub Nitrate + Na Bicarbonate + $H_2O \rightarrow CO_2$ liberate

④ Incompatible of emulsifying agent $\rightarrow$ Here in some cases, phase inversion take place. o/w convert into w/o and w/o convert into o/w.

⑤ Colour stability of dyes $\rightarrow$ Most of the dyes / colour used in pharmaceutical formulation influenced by their ionisation depend on pH of solution.

Example $\rightarrow$

[DEPTH OF BIOLOGY]

 $\rightarrow$ Phenolphthalein $\Rightarrow$ colourless in acidic solution
 $\downarrow$
Red in alkaline mixture.

Therapeutic Incompatibility

Prescriber prescribe to treat a particular effect but this show different nature.

Causes :-

[DEPTH OF BIOLOGY]

- ① overdose
- ② Improper dosage form
- ③ Contraindicated
- ④ Antagonistic drug
- ⑤ synergistic

[DEPTH OF BIOLOGY]